

# South Dakota School of Mines and Technology

## Department of Materials and Metallurgical Engineering

### Darken Plot Slope and the Interaction Coefficient

The Interaction Coefficient the Interaction Parameter are defined as follows:

$$\text{Interaction Coefficient: } e_i^j \equiv \frac{\partial \log f_i}{\partial \text{wt}\% i} \quad (1)$$

$$\text{Interaction Parameter: } \varepsilon_i^i \equiv \frac{\partial \ln f_i}{\partial x_i} \quad (2)$$

The straight line portion of the Darken Plot at the infinite dilution end of the plot may be described as

$$\ln \gamma_i = a + b(1 - x_i)^2 \quad (3)$$

The 1 wt% ss activity coefficient,  $f_i$ , is related to the pure ss activity coefficient in the following way:

$$f_i \equiv \frac{h_i}{h_i^{IDEAL}} = \frac{h_i}{\text{wt}\% i} = \frac{a_i}{a_i^{1\text{wt}\% IDEAL}} = \frac{x_i \gamma_i}{x_i \gamma_i^o} = \frac{\gamma_i}{\gamma_i^o} \quad (4)$$

The equating of  $f_i$  to activities  $a_i$  is most easily accomplished by simply thinking of the activity  $a_i$  vs  $x_i$  plot that is also labeled with an activity  $h_i$  scale. The ratio of activities is the same on either scale.

Combining Eqs [1-3] gives

$$\varepsilon_i^i \equiv \frac{\partial \ln(\gamma_i / \gamma_i^o)}{\partial x_i} = \frac{\partial [\ln \gamma_i - \ln \gamma_i^o]}{\partial x_i} = \frac{\partial [a + b(1 - x_i)^2] - [a]}{\partial x_i} = \frac{b \partial (1 - x_i)^2}{\partial x_i} = -2b(1 - x_i) \quad (5)$$

Since the mole fraction of  $i$  is small compared to unity, the interaction parameter is said to be related to the Darken Plot slope at infinite dilution as follows:

$$\varepsilon_i^i = -2b \quad (6)$$

Converting  $x_i$  to wt %  $i$  and log to ln in the definitions in Eqs [1] and [2] gives

$$\varepsilon_i^i = \frac{\partial \ln f_i}{\partial x_i} = \frac{2.303 \partial \log f_i}{\partial \left[ \frac{\text{wt}\% i / MW_i}{100\% / MW_{Fe}} \right]} = \frac{2.303}{\left[ \frac{MW_{Fe}}{100\% MW_i} \right]} \frac{\partial \log f_i}{\partial \text{wt}\% i} \quad (7)$$

$$\varepsilon_i^i = \frac{2.303 * 100\% MW_i}{MW_{Fe}} e_i^i \quad (8)$$

Therefore the interaction coefficient is

$$e_i^i = -2b \frac{MW_{Fe}}{2.303 * 100\% MW_i} \quad (9)$$